POZNAN UNIVERSITY OF TECHNOLOGY



EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS)

COURSE DESCRIPTION CARD - SYLLABUS

Course name

Corrosion-resistant alloys [S1IMat1>SOnK]

| dr hab. inż. Natalia Makuch-Dzia natalia.makuch@put.poznan.pl | arska prof. PP | | |
|--|-------------------------|-----------------------------------|------------|
| Coordinators | | Lecturers | |
| Number of credit points 2,00 | | | |
| Tutorials 0 | Projects/seminars 0 | 8 | |
| Number of hours Lecture 15 | Laboratory classe 15 | es | Other 0 |
| Form of study full-time | | Requirements elective | |
| Level of study first-cycle | | Course offered in Polish | 1 |
| Area of study (specialization) – | | Profile of study general academic | с |
| Field of study Materials Engineering | | Year/Semester 3/6 | |
| Course | | | |

Prerequisites

Knowledge: Knowledge of engineering materials, basic chemistry. Skills: Logical thinking, independent learning, using the library and the Internet. Social competences: The student is aware of the importance and understands the non-technical aspects and effects of engineering activities, including its impact on the environment.

Course objective

Students receive a complete set of basic information to help understand corrosion phenomena, and are ready to find solutions to corrosive problems with metal materials.

Course-related learning outcomes

Knowledge:

1. the student should have detailed knowledge of k_w10 engineering materials

2. the student should have knowledge of chemistry to understand the structure of elements and chemical compounds; describe the elements of inorganic chemistry, acids, bases, salts, types of reactions, chemical equilibrium, chemical kinetics, electrochemistry, basics of metallurgy - [k_w03]

Skills:

1. the student is able to obtain information from literature, databases and other properly selected sources of materials engineering - $[k_u01]$

2. the student is able to identify and formulate simple engineering tasks of a practical nature, characteristic of materials engineering, including in particular the selection of materials for specific applications; define the working conditions of the materials. - [k_u16]

3. the student is able to select engineering materials, methods of shaping the structure and properties of materials for technical applications depending on the structure, properties and conditions of use - [k_u21]

Social competences:

1. the student understands the need for lifelong learning; can inspire and organize the learning process of other people - $[k_k01]$

2. the student correctly identifies and resolves dilemmas related to the profession - [k_k05]

Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

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Lecture: written credit at the end of the semester (credit if at least 51% of points are obtained: <51% 2 - ndst, 51% -62% 3 - dst, 63% -74% 3.5 - dst +, 75% -84% 4 - db, 85% -94% 4.5 - db +,> 94% 5 - very good). Laboratory: Credit based on a written test and written studies on the program content implemented during exercises. In order to pass the written test and all studies must be passed with a positive mark.

Programme content

Knowing the detailed information about the corrosion processes of various metallic materials and knowing with the basic groups of corrosion-resistant alloys and the characteristics of their corrosion resistance.

Course topics

Lecture:

- 1. Types of corrosion and mechanisms of corrosive destruction of metals and alloys.
- 2. Classification of corrosion-resistant steels. Intergranular corrosion of austenitic corrosion-resistant steels.
- 3. Passivity, passivating metals and non-ferrous alloys.
- 4. Non-ferrous alloys resistant to corrosion.
- 5. Corrosion protection methods.
- 6. Corrosion resistance testing methods.

Laboratory:

- 1. Identification of the corrosion mechanism
- 2. Surface layers and corrosion-resistant coatings
- 3. Corrosion-resistant steels
- 4. Corrosion-resistant nickel alloys
- 5. Corrosion resistance of aluminum alloys. Anodizing of aluminum.

Teaching methods

- 1. Lecture: multimedia presentation
- 2. Laboratory exercises: practical exercises, discussion and teamwork.

Bibliography

Basic

- 1. J. Baszkiewicz, M. Kamiński, "Korozja materiałów", Oficyna wydawnicza PW, Warszawa 2006
- 2. S. Mrowec, T. Werber, Korozja gazowa metali, Wydawnictwo Śląsk, Katowice 1975
- 3. M. Orman, A. Golian, Korozja aluminium i jego stopów, Wydawnictwo Śląsk, Katowice 1963

4. pod red. K. Darowicki, Procesy korozyjne, Politechnika Gdańska, Gdańsk 2008 Additional

- 1. L. A. Dobrzański "Podstawy nauki o materiałach i metaloznawstwo" WNT 2002
- 2. R.K. Tredhewey Corrosion, Longman, 1988

Breakdown of average student's workload

| | Hours | ECTS |
|--|-------|------|
| Total workload | 50 | 2,00 |
| Classes requiring direct contact with the teacher | 35 | 1,00 |
| Student's own work (literature studies, preparation for laboratory classes/ tutorials, preparation for tests/exam, project preparation) | 15 | 1,00 |